

## Combustion Task: "Mushroom compost"

A small power plant in Ireland wishes to burn "mushroom compost", which is a residue product from local mushroom processing industry.

Composition of the compost fuel in %-mass per **dry** fuel:

|                |        |
|----------------|--------|
| C              | 30.2 % |
| O <sub>2</sub> | 24.3%  |
| N <sub>2</sub> | 2.3%   |
| H <sub>2</sub> | 3.5%   |
| S              | 2.7%   |
| A              | 37.0%. |

The higher heating value (HHV) for the **dry** fuel is 12.19 MJ/kg.

The fuel has a high moisture content  $M = 61.8$  % of total mass!

The mushroom compost will be combusted with an air excess of 30% ( $m=1.3$ ).

- How much normalized air is needed ( $m_n^3$ ) per kg of fuel for the combustion process?
- How much flue gases ( $m_n^3$ ) per kg of fuel are produced?
- Calculate the lower heating value (LHV) for the total (wet) fuel.