



MJ2411

SEMINAR ON ENERGY STORAGE

Dr. Justin NW Chiu

justin.chiu@energy.kth.se

+46 (0)87907414

**Department of Energy Technology
Royal Institute of Technology (KTH)
Stockholm, Sweden**



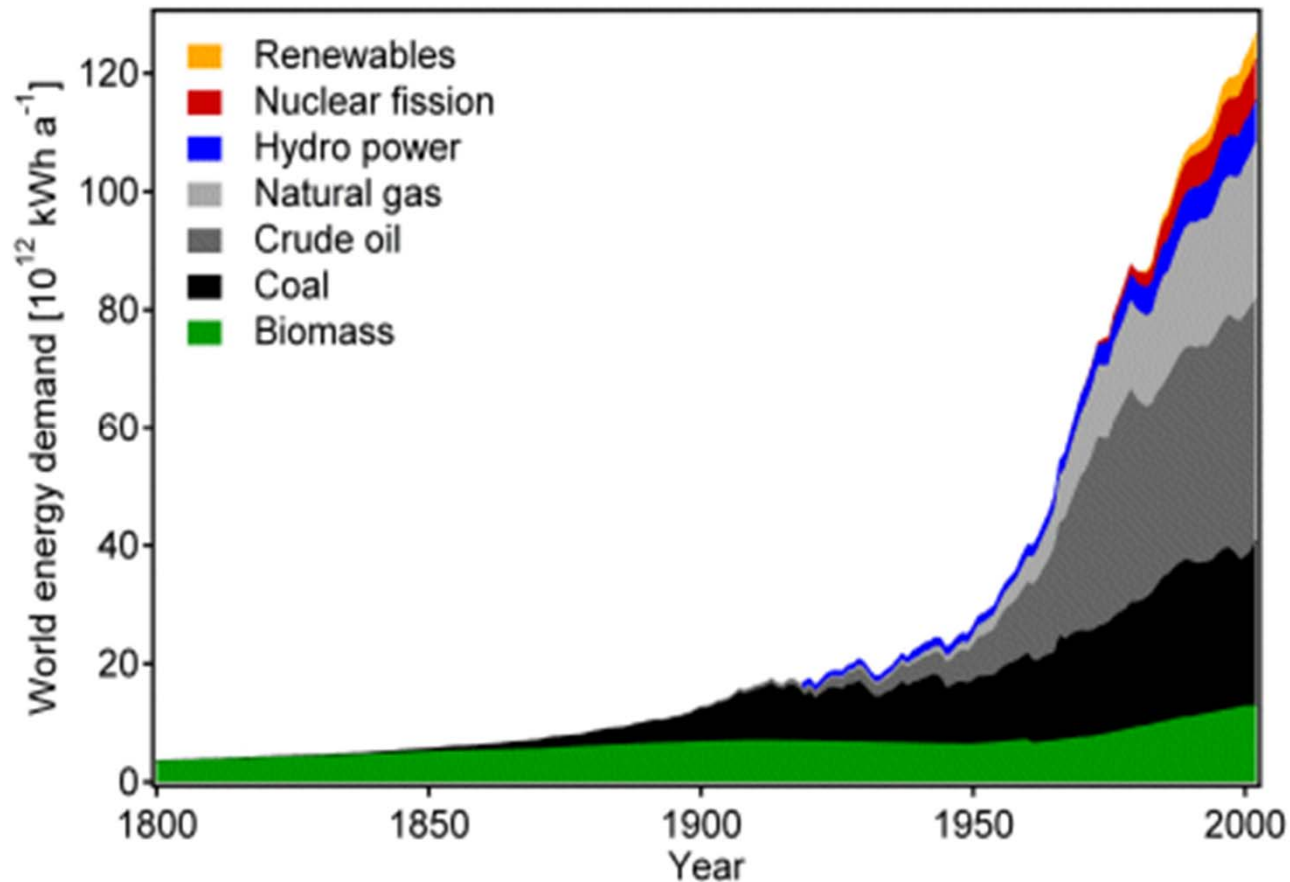
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1. INTRODUCTION TO ENERGY STORAGE

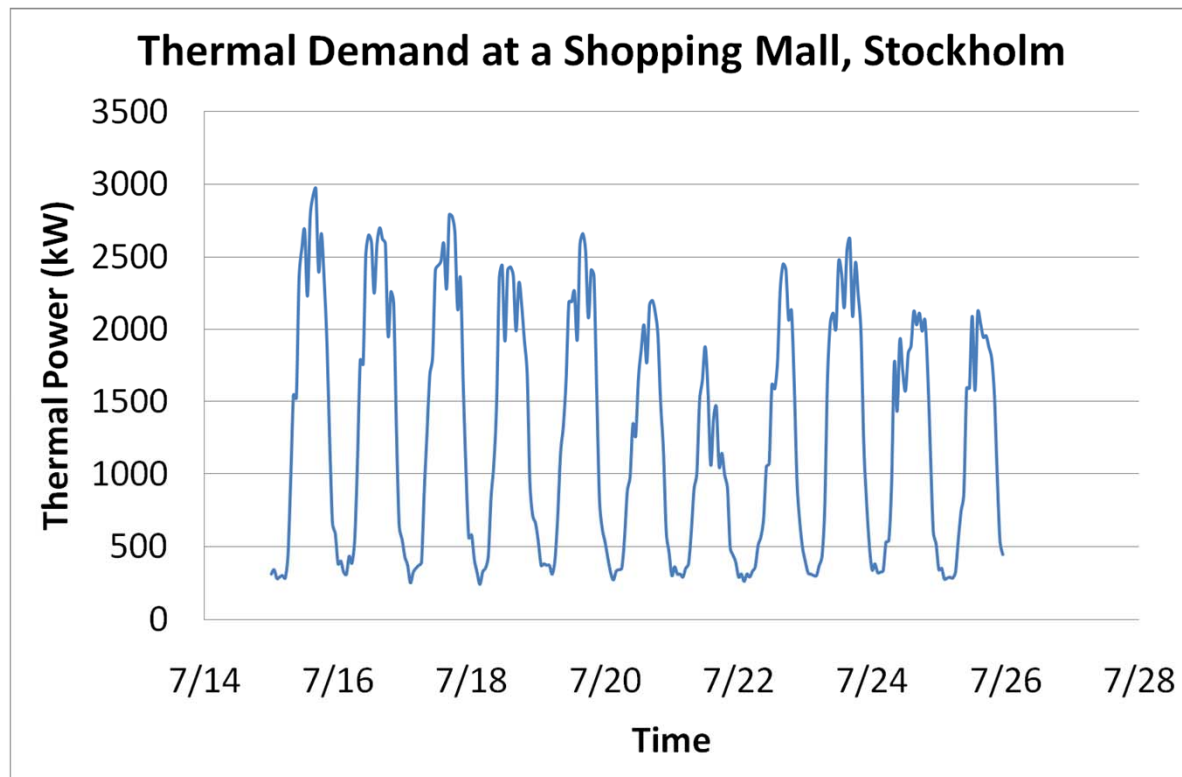
1.1. Increasing Energy Demand



Source: BP Statistical Review of World Energy.

1.2. Mismatch in Time and Space

- Intermittent energy resources
- Fluctuating energy use



Source: Fortum AB.



1.3. Method: Peak Shaving/ Load Leveling

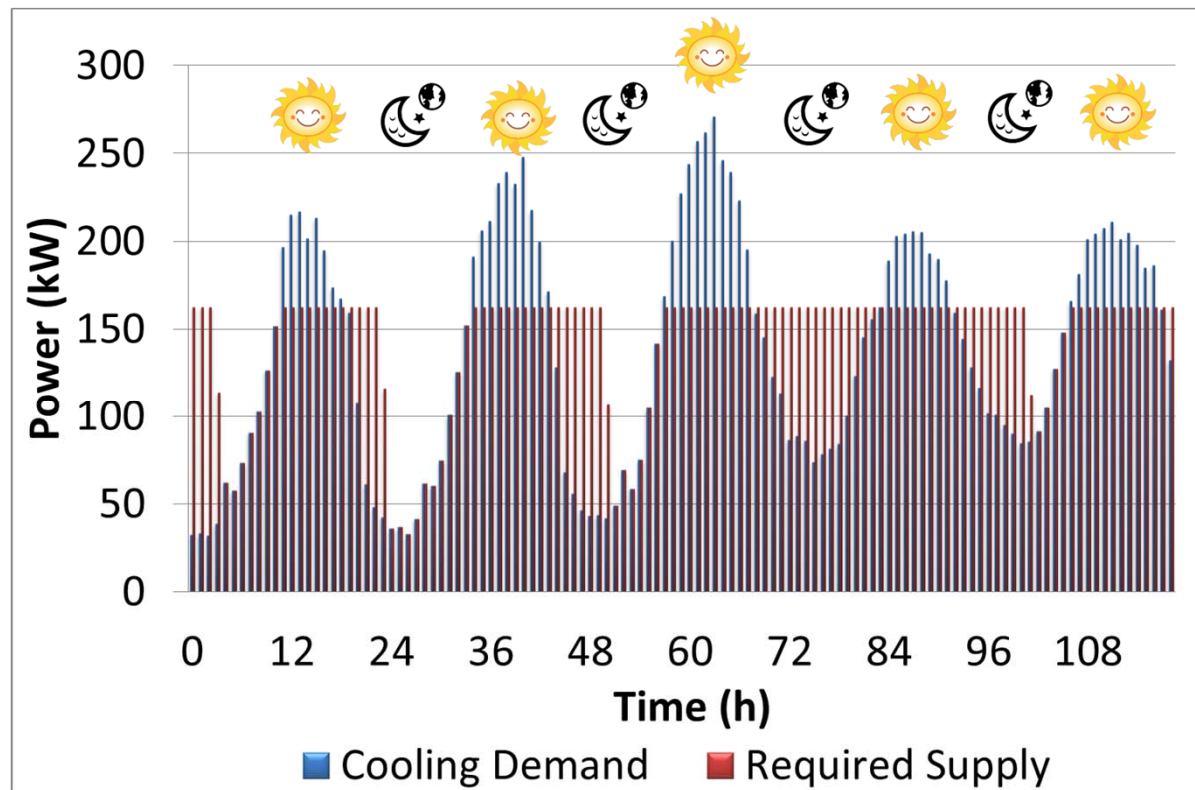
Peak shave → additional power

Load shift → off-peak storage for peak use

Hourly
Storage

Daily
Storage

Seasonal
Storage





1.4. Benefits with Energy Storage

- Energy Arbitrage → Buy and store low-cost energy, sell when price is high
- Generation Capacity Deferral → reduce peak generation, maintain nominal efficiency
- Transmission and Distribution Deferral → alleviate T&D investment
- Ancillary Service → balance supply and demand, reserve capacity
- Reduction of Renewable Curtailment → due to minimum load, ramping, grid limit



2. CATEGORIES OF ENERGY STORAGE



2. Categories of Energy Storage

Mechanical Storage

- Compressed Air
- Fly Wheel
- Pumped Hydro

Electrochemical Storage

- Batteries

Electromagnetic Storage

- Capacitor
- Magnetic storage

Chemical Storage

- Hydrogen
- Hydrocarbons

Thermal Storage

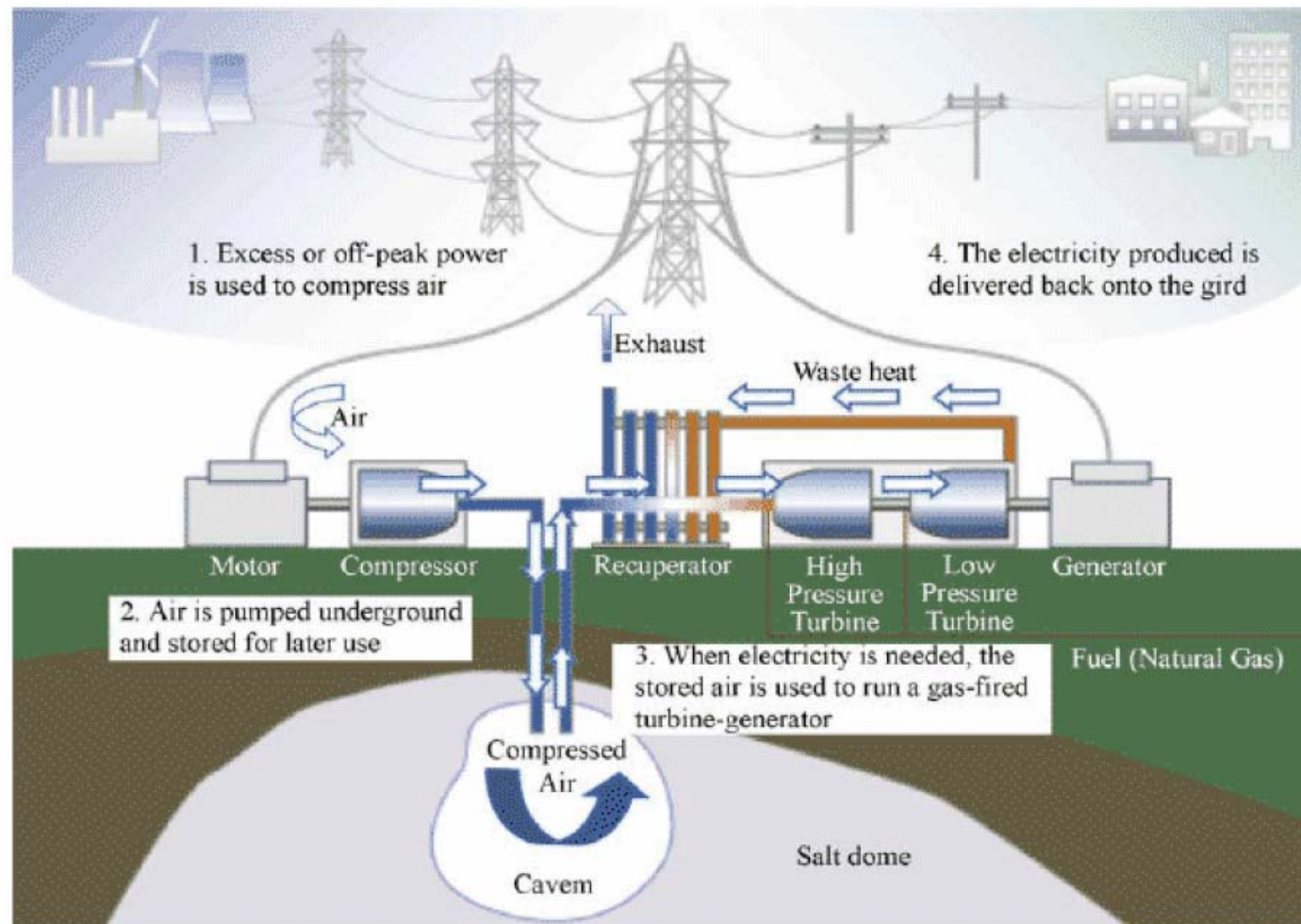
- Phase change Material

Kinetic Energy

Potential Energy



2.1. Compressed Air Energy Storage





2.1. Compressed Air Energy Storage

Derive reversible adiabatic (isentropic) compression work required to compress ideal gas.

$$PV^n = \text{constant}$$

$$\begin{aligned} W &= - \int_1^2 P \cdot dV = - \int_1^2 cte \cdot \frac{dV}{V^n} = - \frac{cte \cdot V_2^{1-n} - cte \cdot V_1^{1-n}}{1-n} \\ &= \frac{P_2 \cdot V_2^n \cdot V_2^{1-n} - P_1 \cdot V_1^n \cdot V_1^{1-n}}{n-1} = \frac{P_2 \cdot V_2 - P_1 \cdot V_1}{n-1} \quad \text{or} \quad \frac{P_1 \cdot V_1}{n-1} \left[\left(\frac{P_2}{P_1} \right)^{\frac{n-1}{n}} - 1 \right] \end{aligned}$$

or

$$= \frac{m \cdot r \cdot T_2 - m \cdot r \cdot T_1}{n-1}$$

$$\text{with } \frac{T_2}{T_1} = \left(\frac{P_2}{P_1} \right)^{\frac{n-1}{n}} = \left(\frac{V_2}{V_1} \right)^{1-n}$$



2.2. Flywheel

Stored energy:

$$KE = \frac{1}{2} I \omega^2 = \frac{1}{2} kmR^2 \omega^2$$

ω = rotational velocity

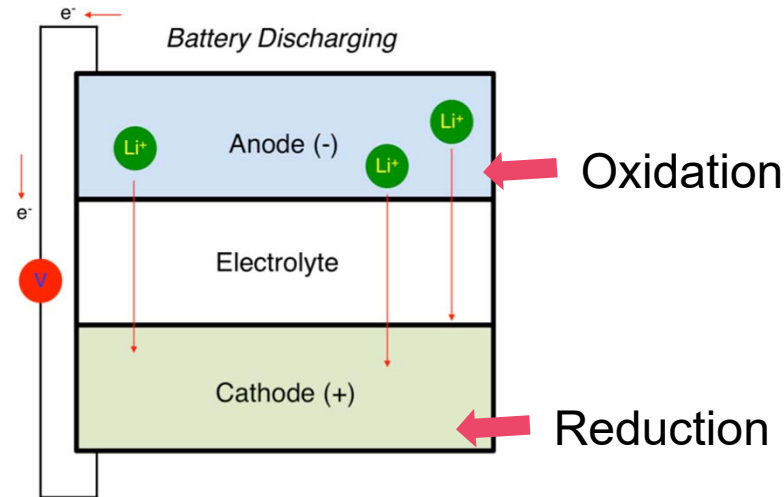
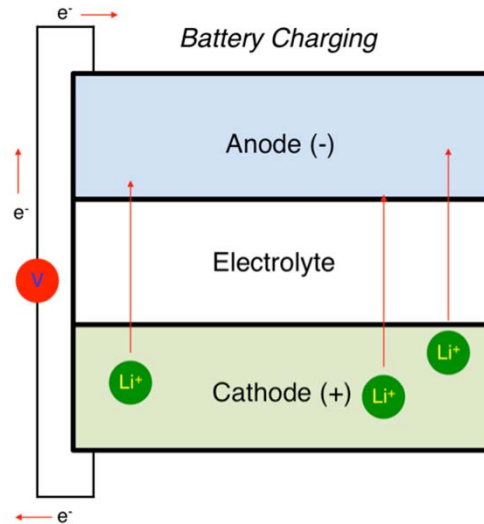
I = moment of inertia (ability of an object to resist changes in its rotational velocity)

$$= kmR^2$$

k : inertial constant depends on shape; m : mass; R : radius

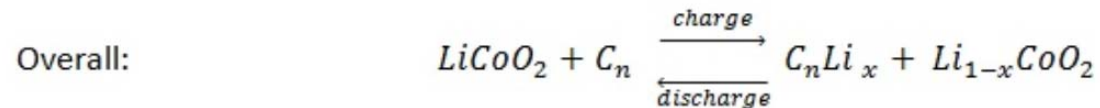
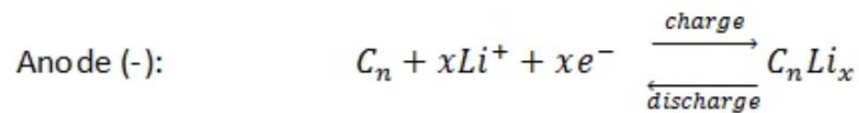
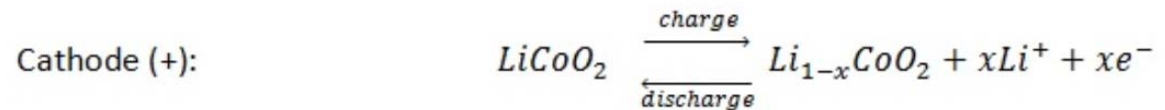
Wheel loaded at rim (bike tire); $k = 1$: solid disk of uniform thickness; $k = 1/2$ solid sphere; $k = 2/5$; spherical shell; $k = 2/3$; thin rectangular rod; $k = 1/2$

2.3. Batteries



Storage Density

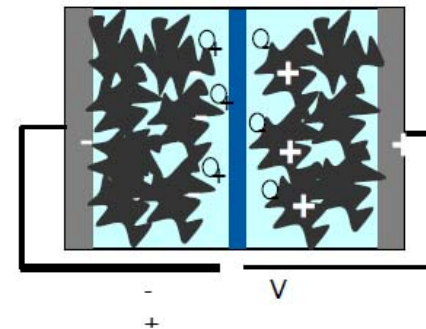
- 30- 250 Wh/kg
- 100- 650 Wh/L



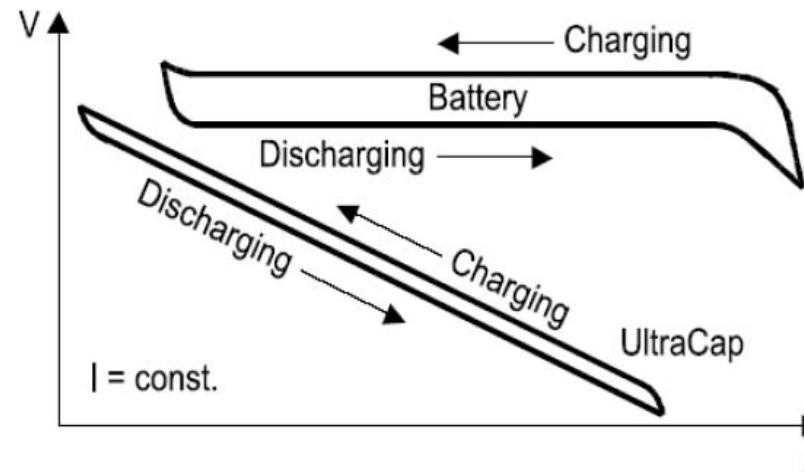
2.4. Capacitors

Function double layer capacitor (supercap)

- electrostatic mechanism
- no chemical conversion involved
- Fast and highly reversible processes

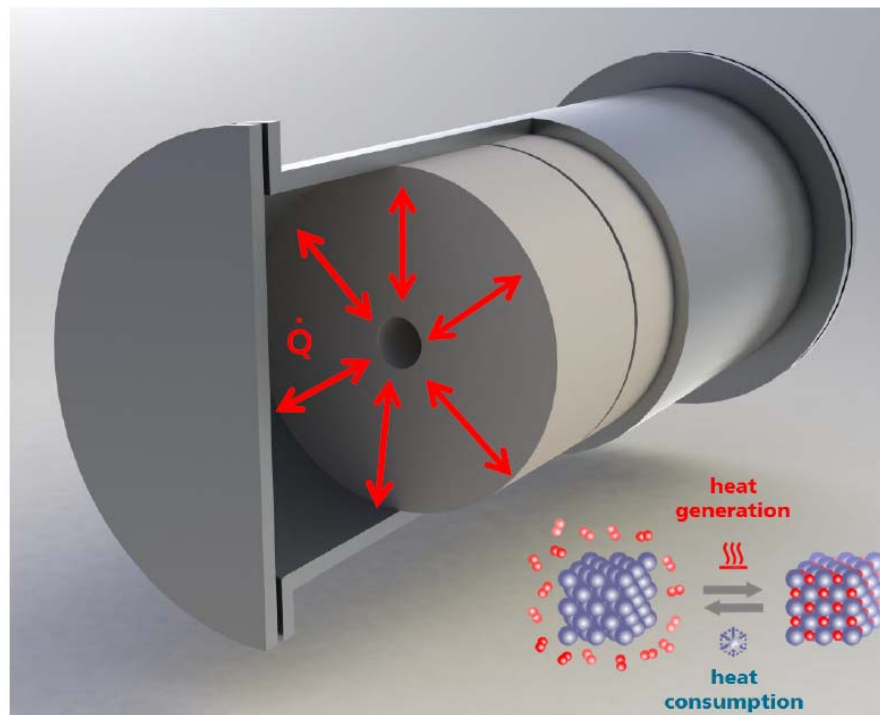


- high cycle life
- high power density
- linear potential increase during charging



2.5. Hydrogen Storage

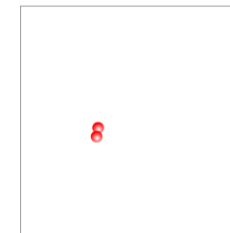
Metal Hydride Storage Systems



- mechanical integrity
- gas permeability
- thermal conductivity

$\lambda \sim 10 \text{ to } 20 \text{ W/(mK)}$
 $\rightarrow$ sufficient heat transfer

H₂ (gas)
 1 bar, 20°C



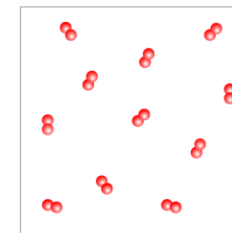
$5.6 \times 10^{19} \text{ H/cm}^3$

H₂ (compressed gas)
 700 bar, 20°C



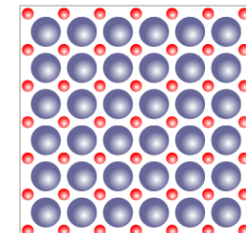
$2.3 \times 10^{22} \text{ H/cm}^3$

H₂ (liquid)
 1 bar, -253°C



$4.2 \times 10^{22} \text{ H/cm}^3$

Hydride (solid)
 20 bar, 20°C

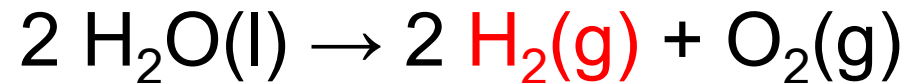


$9.3 \times 10^{22} \text{ H/cm}^3$



2.6. Power to Gas

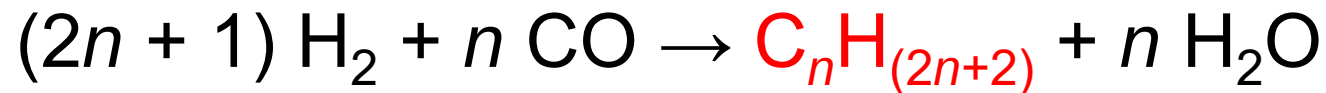
Electrolysis of Water



Reverse Water Gas Shift Reaction



Fischer Tropsch Process





2.7. Thermal Energy Storage

☐ Sensible Heat Storage

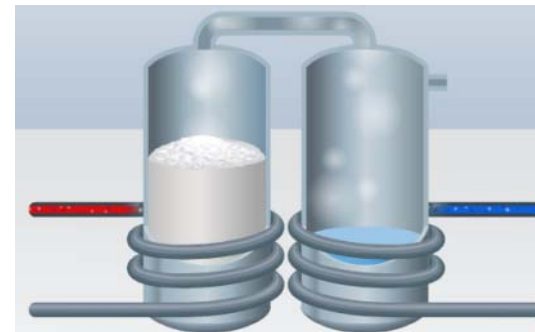
- Wide range of materials

☐ Latent Heat Storage- **Phase Change Material**

- High storage density
- Small temperature swing

☐ Thermo-Chemical Storage

- Long term storage



Source: P. Widmer, A. Castell, Climatewell



2.7. Thermal Energy Storage

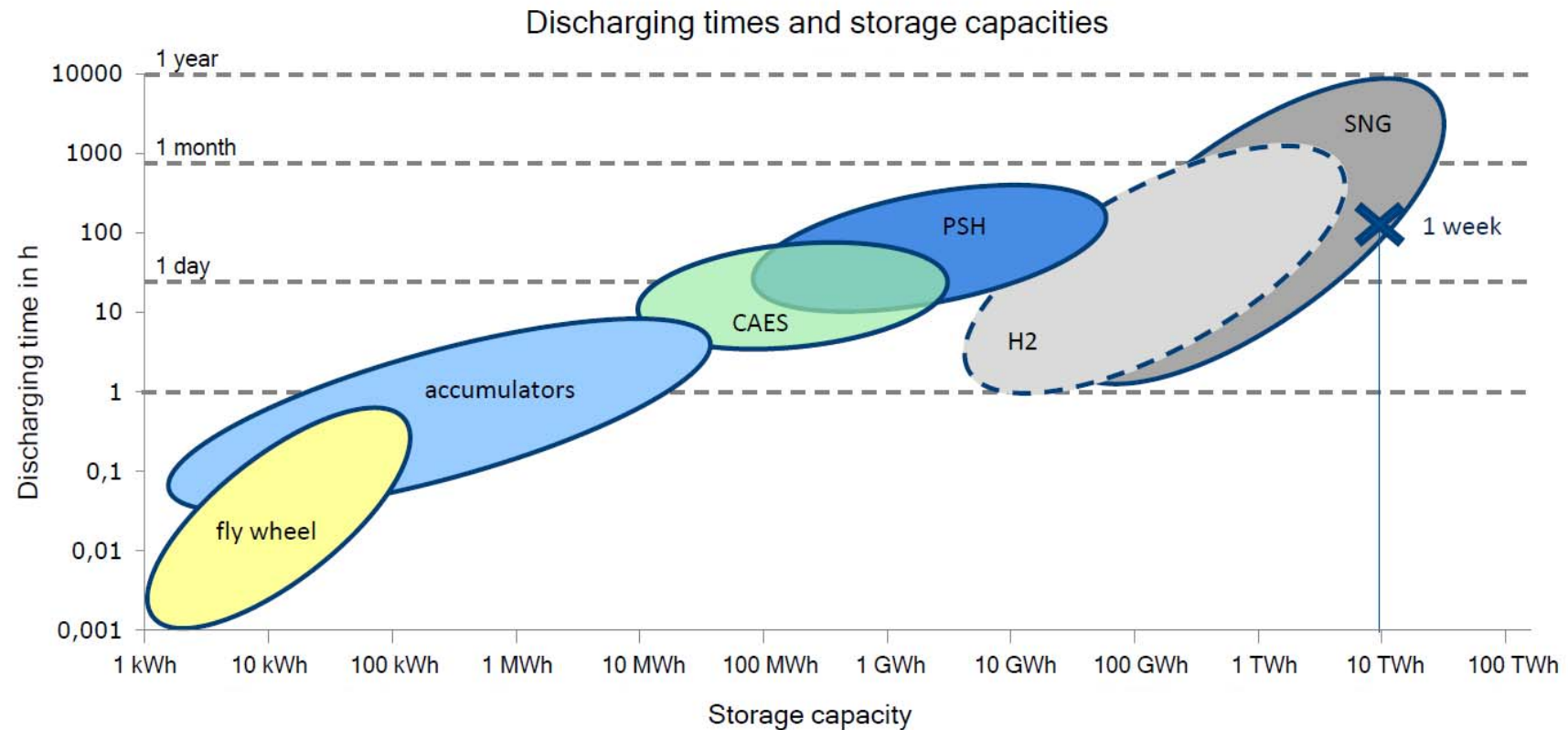
Amount of energy stored (Q) over temperature range $[T_1; T_2]$ with λ specific latent heat, h specific enthalpy, m mass, and c_p specific heat capacity:

$$Q_{pcm} = \int_{T_1}^{T_{PC}} m \cdot c_{p\ sol}(T) \cdot dT + m \cdot \lambda + \int_{T_{PC}}^{T_2} m \cdot c_{p\ liq}(T) \cdot dT$$

In most cases, phase transition is accompanied with temperature change, hence:

$$Q_{pcm} = \int_{T_1}^{T_2} m \cdot dh_{pcm}(T) = \int_{T_1}^{T_2} m \cdot c_{p\ pcm}(T) \cdot dT$$

2.8. Comparison of Storage Technologies





3. EXERCISES



Exercise 1.

$T_{ini} = 50^\circ\text{C}$; $T_{fin} = 70^\circ\text{C}$. What is the storage density for water? What is the storage density for S58 PCM? With packing factor (PF) of 0.8 for latent heat storage. What is the volume uptake ratio between the two storages?

Water: $c_p = 4.18 \text{ kJ/kg-K}$ and $\rho = 1 \text{ kg/L}$

S58 PCM: $c_p = 2.55 \text{ kJ/kg-K}$, $\lambda = 145 \text{ kJ/kg}$ and $\rho = 1.5 \text{ kg/L}$

- For water: $q_{\text{water}} = 4.18 * (70-50) = 83.6 \text{ kJ/kg}$ or 83.6 kJ/L

- For PCM S58: $q_{\text{PCM}} = 2.55 * 20 + 145 = 196 \text{ kJ/kg}$ or 294 kJ/L

- $V_{\text{pcm}}/V_{\text{water}} = q_{\text{water}}/(q_{\text{pcm}} * \text{PF}) = 0.355$ in volume uptake

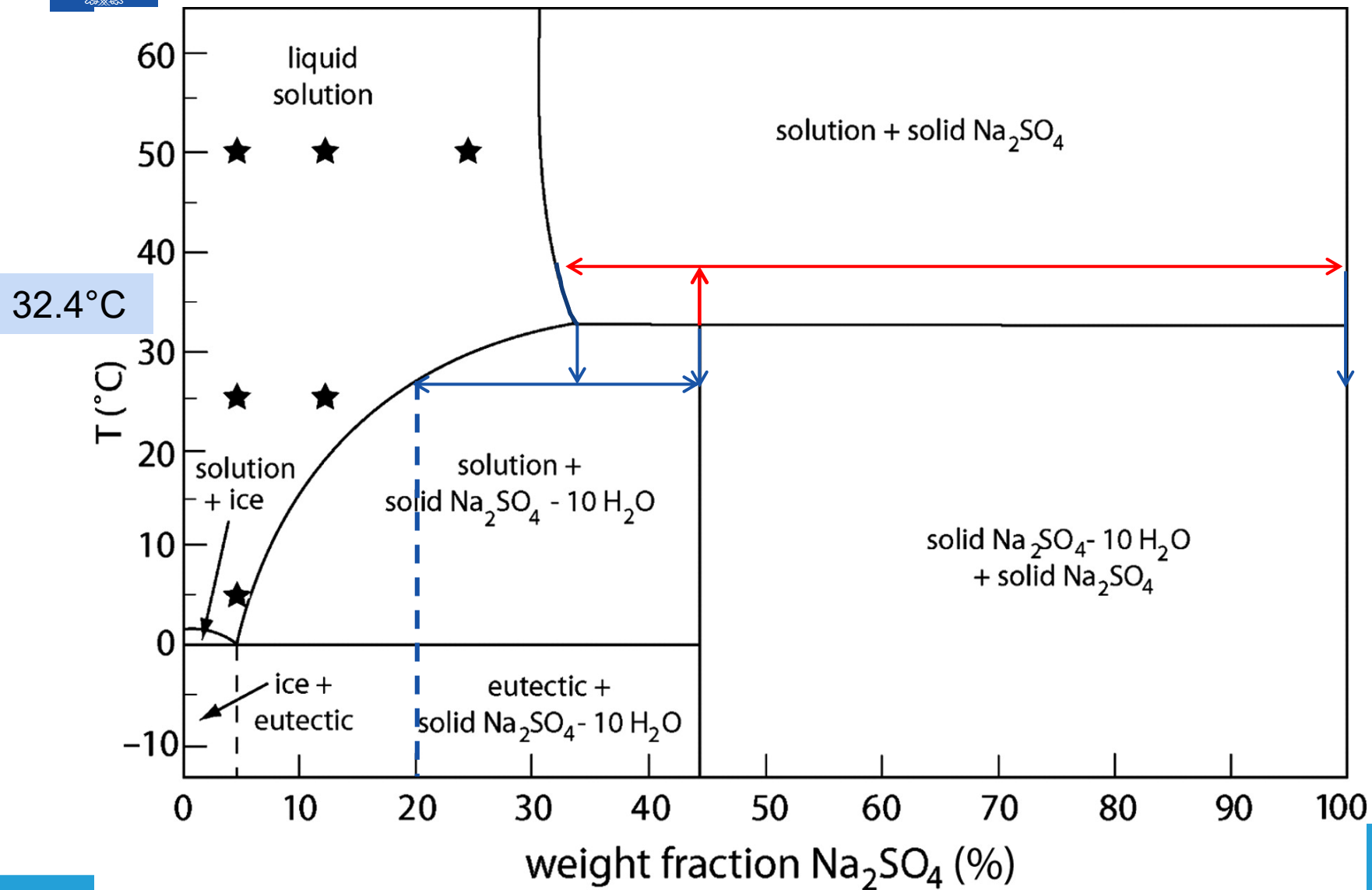


Discussion

- In reality $\rho_{\text{liquid}} \neq \rho_{\text{solid}}$, how does this affect the design?
 - volume expansion is to be taken into account
 - tank has to cope with pressure increase
 - consider if buoyancy mechanism affects the performance

- How is packing factor (PF) linked to the storage performance?
 - higher PF will increase the energy density, but lowers the thermal power rate
 - higher thermal power rate will lower the PF which lowers the energy density

Exercise 2.





Exercise 2.

- After cooling molten Glauber's Salt ($\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$) from 32.4°C to 26°C , how much Glauber's Salt is left?
- Hint: use the lever rule calculation on the previously shown phase diagram

$$\blacksquare m_{GS} \cdot (44.1\% - 33.2\%) = m_{\text{Solution}} \cdot (33.2\% - 20\%)$$

$$\rightarrow m_{GS} \times 82.6\% = m_{\text{Solution}}$$

or

$$\blacksquare X_{GS} = \frac{m_{GS}}{m_{\text{total}}} = \frac{33.2\% - 20\%}{44.1\% - 20\%} = 54.8\%$$



Discussion

- What are the negative impacts of phase segregation?
 - reduced storage capacity
 - alteration of phase change temperature

4. GENERAL DISCUSSIONS



4.1. Design Questions

1. Which storage system to use?
2. What storage characteristics?
 - Storage capacity
 - Volume update, Weight limitation
 - Peak power rate
 - Storage energy loss rate
 - Cyclability of the storage
 - Cycle efficiency
 - Depth of charge/discharge
 - Ramp up time, etc.
3. How can storage fit and operate with existing systems?
4. What are the environmental, social, economic benefits and drawbacks?



**Thank you for your attention.
Question?**

justin.chiu@energy.kth.se

